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Matthias Zeller,^a Allen D. Hunter,^a* Paul Sampson^b and Nataliya Chumachenko^b

^aDepartment of Chemistry, Youngstown State University, One University Plaza, Youngstown, OH 44555-3663, USA, and ^bDepartment of Chemistry, Kent State University, PO Box 5190, Kent, OH 44242-0001, USA

Correspondence e-mail: adhunter@ysu.edu

Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.002 Å R factor = 0.035 wR factor = 0.097 Data-to-parameter ratio = 14.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(4aS*R,7RS,*9aS*R*)-7,9a-Dimethyl-5,5-dioxo-1,2,4a,6,7,9a-hexahydro-8-oxa-5λ⁶-thia-

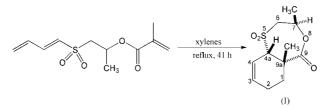
1,4-benzocyclohepten-9-one

The title compound, $C_{11}H_{16}O_4S$, crystallizes as a racemic mixture in space group $P_{2_1/n}$. The seven-membered ring exhibits a chair conformation and the methyl substituents point away from each other.

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Comment

As detailed in the first article in this series (Zeller *et al.*, 2004), two of us (NC and PS) have been exploring the potential utility of sulfone-based tethers in intramolecular Diels–Alder cycloaddition reactions. This has resulted in the preparation of a series of bicyclic β -acyloxy sulfone cycloadducts, several of which have been subjected to analysis using single-crystal Xray diffraction. In the present paper, the solid-state structure of the title cycloadduct, (I), as established by single-crystal Xray diffraction methods, is described.



Compound (I) crystallizes in space group $P_{1/n}$, with Z = 4, as a racemic mixture of both enantiomers. The six-membered ring exhibits the half-chair conformation expected for cyclohexenes, and the angles at the unsaturated C atoms are $123.93 (12)^{\circ}$ (for C4–C3–C2) and $124.05 (12)^{\circ}$ (for C3– C4–C4A; Fig. 1). The seven-membered ring displays a chairlike conformation. The methyl group on atom C7 is located in the equatorial position, pointing away from the methyl substituent on the bridgehead atom C9A.

Experimental

Compound (I) was isolated in 48% yield as the major diastereomer from an intramolecular Diels–Alder cycloaddition reaction of 1-[(E)buta-1,3-dienylsulfonyl]propan-2-yl methacrylate in refluxing xylenes. The crude product yielded crystals of (I) suitable for X-ray diffraction analysis *via* crystallization from ethyl acetate/hexane (1:2) with slow cooling.

Crystal data $C_{11}H_{16}O_4S$ $D_x = 1.430 \text{ Mg m}^{-3}$ $M_r = 244.30$ Mo $K\alpha$ radiation Monoclinic, $P2_1/n$ Cell parameters from 9433 a = 10.0232 (10) Åreflections b = 10.6255 (10) Å $\theta=2.2{-}28.3^\circ$ $\mu = 0.28~\mathrm{mm}^{-1}$ c = 11.6726 (11) Å $\beta = 114.062 \ (2)^{\circ}$ T = 100 (2) K $V = 1135.13 (19) \text{ Å}^3$ Block, colorless Z = 4 $0.4 \times 0.2 \times 0.2$ mm

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organic papers

Data collection

Bruker SMART APEX CCD diffractometer ω scans Absorption correction: multi-scan (*SADABS* in *SAINT-Plus*; Bruker, 1997–1999) *T*_{min} = 0.735, *T*_{max} = 0.945 11460 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.035$ $wR(F^2) = 0.097$ S = 1.102829 reflections 193 parameters Only coordinates of H atoms refined 2829 independent reflections 2729 reflections with $I > 2\sigma(I)$ $R_{int} = 0.024$ $\theta_{max} = 28.3^{\circ}$ $h = -13 \rightarrow 13$ $k = -14 \rightarrow 14$

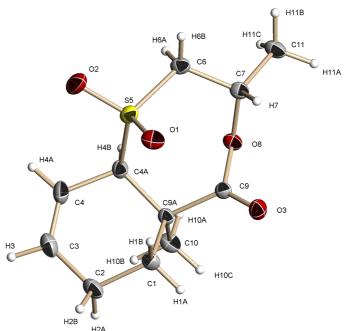
$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0521P)^2 \\ &+ 0.5113P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} &= 0.002 \\ \Delta\rho_{\text{max}} &= 0.44 \text{ e } \text{\AA}^{-3} \\ \Delta\rho_{\text{min}} &= -0.32 \text{ e } \text{\AA}^{-3} \end{split}$$

 $l = -15 \rightarrow 15$

All H atoms were positioned geometrically. Their coordinates were then refined freely, and their U_{iso} values were defined as 1.2 or $1.5U_{eq}$ of the parent C atom. The s.u. values of the cell parameters are taken from the software, recognizing that the values are unreasonably small (Herbstein, 2000).

Data collection: *SMART* (Bruker, 1997–2000); cell refinement: *SAINT-Plus* (Bruker, 1997–1999); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1990); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXL*97.

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The molecular structure, showing 50% probability displacement ellipsoids.

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